# LIGNANS AND OTHER CONSTITUENTS FROM CARISSA EDULIS\*

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Key Word Index—Carissa edulis; Apocynaceae; lignans; lignan carissanol; carisson; aromatic compounds; 4-hydroxy-(3-hydroxypropionyl)-benzene.

Abstract—The methanolic extract from the root of Carissa edulis contains ca 5% lignans, the main compounds among them being (-)-nortrachelogenin, carinol and the hitherto unknown carissanol.

#### INTRODUCTION

Various plants of the genus Carissa are used in India and other regions of Asia as medicinal agents. [1]. In West Africa, the root of C. edulis is reported as a remedy against various diseases [2, 3]. The presence of cardioactive substances has been demonstrated in some Asian Carissa species [4, 5] and in consequence the Indian C. carandas has been studied phytochemically in more detail [6-10]. Very little is known about the constituents of C. edulis: a French group isolated the inosit derivative quebrachitol (1) from the twigs [11] and, in a general screening test for digitalis-type glycosides, extracts were found to be negative [12].

#### RESULTS AND DISCUSSION

The methanolic extract from the root of *C. edulis* was concentrated and partitioned between (a) petrol, (b) tetrachloromethane, (c) diethyl ether, (d) ethyl acetate and (e) water. The petrol phase contained a mixture of pentacyclic triterpenes and o-hydroxyacetophenone (2); in the tetrachloromethane phase we found traces of catalponol (3) [13, 14] and a mixture of sesquiterpenes, mainly carisson (4), which was first isolated from *C. lanceolata* in 1954 [15] and later in two other *Carissa* species [1, 16]. Chromatography of the ether phase on Si gel yielded, besides small amounts of the aromatic compounds 5–9, a mixture of lignans (90% of the ether phase) which were further separated chromatographically into the three major lignans 10–12 and the three minor lignans 13–15.

Compounds 10 and 11 were identified, on the basis of their physicochemical data and the spectra of their acetylation products, as (-)-nortrachelogenin (10) [17, 18] and carinol (11) [19], respectively.

The structure of 12 was deduced from its <sup>1</sup>H, <sup>13</sup>C NMR and mass spectra and the corresponding spectra of its acetyl derivatives. In agreement with the lactol group in 12 are the observations that 12 (a) in solution exists, according to <sup>1</sup>H and <sup>13</sup>C NMR spectra, as a mixture of epimers, (b) forms two epimeric O-alkyl derivatives on treatment with alcohol-H<sup>+</sup>, and (c) is converted into 11 by reduction with lithium aluminium hydride in diethyl ether. On the other hand 12, as well as 11, can be prepared by careful reduction of 10 with lithium aluminium hydride in tetrahydrofuran. This latter reaction also determines the absolute configuration of 12.

In the same manner as for 10 and 11, the structures of the minor lignans were identified as the already known compounds (+)-lariciresinol (13) [20], (-)-secoisolariciresinol (14) [21] and (-)-olivil (15) [20].

The lignans altogether constitute ca 5% of the total methanolic extract and the three major compounds 10-12 ca 1% each. Compound 12 represents a new lignan of the comparatively rare lactol-type; it consequently was named carissanol. (-)-Nortrachelogenin has been found only once as a constituent of *Trachelospermum asiaticum* (Apocynaceae), a plant used in China against various diseases [24]. (+)-Nortrachelogenin, the enantiomer of

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10, which was isolated from some *Wikstroemia* species (Thymelaeaceae) [25, 26, 28] has been proved to possess antitumor activity [26]. Up to now, no biological activities have been reported for (-)-nortrachelogenin (10) or carinol (11).

#### **EXPERIMENTAL**

General procedures. If not stated otherwise, all general procedures and instruments were as described in ref. [22]. MS were run at 70 eV. For TLC we used 0.25 mm Si gel N/UV<sub>254</sub> (Macherey-Nagel) and the solvent system CHCl<sub>3</sub>-MeOH (9:1); detection by UV at 254 nm and by anisaldehyde reagent No. 15 according to Stahl [23].

Plant material. Roots of C. edulis Vahl were collected near Legon, Ghana, in March 1979 and identified by Mr. A. A. Enti. A herbarium specimen is deposited in our collection under No. 79/1.

Extraction procedure. Powdered dried roots (2.5 kg) were extracted with 151. MeOH in a Soxhlet apparatus. On evaporation this extract yielded 155 g residue, which was redissolved in 31. MeOH- $\rm H_2O$  (1:1) and then treated successively with 8  $\times$  250 ml petrol (upon evaporation: 35 g extract A), 8  $\times$  250 ml CCl<sub>4</sub> (8 g extract B), 20  $\times$  250 ml Et<sub>2</sub>O (11 g extract C) and 20  $\times$  500 ml EtOAc (13 g extract D).

Extract A was separated over 300 g Si gel (column operated at ca 30 psi) into eight fractions (A1-A8) using cyclohexane with increasing amounts of EtOAc as eluent.

o-Hydroxyacetophenone (2). Rechromatography of A3 over 250 g Si gel with petrol-Me<sub>2</sub>CO (19:1) gave 1.2 g 2.

Catalponol (3). Extract B contained mainly carisson (4) and other sesquiterpenes. Besides these, repeated chromatography over Si gel with different solvent systems afforded ca 4 mg of a viscous oil,  $[\alpha]_2^{10} + 16^{\circ}$  (CHCl<sub>3</sub>; c 0.31), physicochemical data identical with 3 [13, 14]. Extract C was separated over 200 g Si gel (column operated at ca 30 psi) into five fractions (C1-C5) using CHCl<sub>3</sub> with increasing amounts of MeOH as eluent. C1 by rechromatography over 250 g Sephadex LH 20 with MeOH-CHCl<sub>3</sub> (7:3) gave three fractions (C1.1-C1.3). C1.2 was further separated over 160 g Si gel (petrol-EtOAc, 3:2) into nine fractions (C1.2.1-C1.2.9).

Vanillin (5) and coniferaldehyde (7). On evaporation, from C1.2.4 crystallized 10 mg 5 and from C1.2.5 14 mg 7.

(-)-Nortrachelogenin (10). On evaporation, C1.2.7 afforded

275 mg of an amorphous powder,  $[\alpha]_D^{20} - 25^\circ$  (EtOH; c 1.5), physicochemical data identical with 10 [17, 18]; <sup>13</sup>C NMR: Table 2.

Scopoletin (8). Purification of C1.3 over 6 g Si gel (toluene-MeOH, 19:1) yielded 3 mg 8 [29].

'Isofraxidin (9). Rechromatography of C2 over 60 g Si gel (CHCl<sub>3</sub>-EtOAc, 3:2) gave 6.5 mg 9 [27].

C3 was separated over 400 g Si gel using CHCl<sub>3</sub>-EtOAc (1:1) into four fractions (C3.1-C3.4). Further chromatography of C3.2 over 5 g Si gel using CHCl<sub>3</sub>-EtOAc-MeOH (5:4:1) gave three fractions (C3.2.1-C3.2.3).

4-Hydroxy-(3-hydroxypropionyl)-benzene (6). Evaporation of C3.2.1 yielded 1.5 mg 6 as a colourless powder; MS: m/z (int.  $\geq 5\%$ ) 166.0638 C<sub>9</sub>H<sub>10</sub>O<sub>3</sub> (18, M<sup>+</sup>), 148 (6), 123 (11), 122 (8), 121 (100), 93 (21), 65 (14); UV  $\lambda_{\max}^{\text{EtOH}}$  nm (log  $\varepsilon$ ): 220 (4.15), 273sh, 279 (4.32); IR  $\nu_{\max}^{\text{CHCl}_3}$  cm<sup>-1</sup>: 3520, 1602; <sup>1</sup>H NMR (Me<sub>2</sub>CO-d<sub>6</sub>):  $\delta$ 9.22 (1H<sub>ex</sub>, s, ar-OH), 7.93 and 6.93 (4H, AA'BB', ar-H), 3.92 (2H, dt,  $J_1 \approx J_2 \approx 6$  Hz, CH<sub>2</sub>-CH<sub>2</sub>-OH), 3.60 (1H<sub>ex</sub>, t,  $J_2 \approx 6$  Hz, OH), 3.14 (2H, t,  $J_3 \approx 6$  Hz, CO-CH<sub>2</sub>-CH<sub>2</sub>-OH).

(+)-Lariciresinol (13). On concn, C3.2.3 gave 2 mg colourless crystals [mp  $166-169^{\circ}$ , [ $\alpha$ ] $_{20}^{20}+10^{\circ}$  (EtOH; c 0.2)], data identical with 13 [20]; acetylation (Ac<sub>2</sub>O-pyridine) converted 13 into the triacetyl derivative, whose spectra were identical with published data [30].

Carissanol (12). Rechromatography of C3.4 over 150 g Si gel with the lower phase of CHCl<sub>3</sub>–MeOH–H<sub>2</sub>O (18:1:1) afforded 250 mg 12 as an amorphous powder;  $[\alpha]_D^{20}$  – 14° (EtOH; c 0.7); C<sub>20</sub>H<sub>24</sub>O<sub>7</sub> (376.4). (Found: C, 63.92; H, 6.26. Calc.: C, 63.82; H, 6.43%.) MS: m/z (int. ≥ 5%) 376.1538 C<sub>20</sub>H<sub>24</sub>O<sub>7</sub> (9, M<sup>+</sup>), 193 (8), 175 (8), 164 (5), 163 (13), 150 (5), 139 (6), 138 (36), 137 (100), 131 (8), 124 (7), 123 (7), 122 (11), 107 (5), 94 (9), 91 (5), 77 (9), 65 (6), 55 (11), 51 (7); UV  $\lambda_{\text{max}}^{\text{EiOH}}$  nm (log ε): 228 (4.05), 282 (3.72); IR  $\nu_{\text{max}}^{\text{KBr}}$  cm<sup>-1</sup>: 3430, 1605, 1515; <sup>1</sup>H NMR: Table 1; <sup>13</sup>C NMR: Table 2.

C4 was separated over 250 g Sephadex LH 20 with MeOH-CHCl<sub>3</sub> (7:3) into five fractions (C4.1-C4.5).

(-)-Secoisolariciresinol (14). Rechromatography of C4.3 over 60 g Si gel using CHCl<sub>3</sub>-MeOH-EtOAc (45:2:3) gave five fractions (C4.3.1-C4.3.5). C4.3.3, upon purification over 10 g Si gel (toluene-MeOH, 9:1), afforded 10 mg crystals [mp 113-114°,  $[\alpha]_D^{20}$  - 28° (EtOH; c 0.33)] physicochemical data identical with 14 [21].

(-)-Olivil (15). Repeated chromatography of C4.4 over firstly 60 g Si gel (CHCl<sub>3</sub>-MeOH-EtOAc, 45:2:3) and then 3 g Si gel (CHCl<sub>3</sub>-MeOH, 19:1) gave 4 mg colourless crystals

Table 1. <sup>1</sup>H NMR data of lignans (in acetone-d<sub>6</sub>)

15	7.14 (d, J = 2)	6.96 $(d, J = 2)$ 6.73 $(d, J = 8)$	6.75 (d, J = 8)	6.88 $(dd, J_1 = 8, J_2 = 2)$	6.77 (dd, $J_1 = 8$ , $J_2 = 2$ ) 4.72 (d, $J = 7.5$ )	2.95 (d, J = 14)	3.05 (d, $J = 14$ ) 2.31 (ddd, $J_1 \approx J_2 \approx J_3 \approx 6-7$ )	1		3.95-3.73 (3H)*		3.60 (d, J = 9)	3.83 (6H, s)	7.39	7.48 3.63	3.9	1
14	674 OH 4 1= 2)	6.71 (2H, d, $J = 8$ )		6.61 (2H, $dd$ , $J_1 = 8$ , $J_2 = 2$ )	2.72 $(dd, J_1 = 14, J_2 = 7.5)$ 2.64 $(dd, J_1 = 14, J_2 = 6.5)$	2.72 $(dd, J_1 = 14, J_2 = 7.5)$	$2.64 \ (dd, J_1 = 14, J_2 = 6.5)$	1.92 (2H, m)	$3.69 (ddd, J_1 = 11, J_2 \approx J_3 \approx 4)$	3.54 (ddd, $J_1 = 11, J_2 \approx J_3 \approx 5$ )	$3.69 (ddd, J_1 = 11, J_2 \approx J_3 \approx 4)$	3.54 (ddd, $J_1 = 11, J_2 \approx J_3 \approx 5$ )	3.77 (6H, s)	7 30 (2H s)	(c (tra) ) (c)   -	4.14 (2H, dd, $J_1 \approx J_2 \approx 4-5$ )	
13	6.95 (d, J = 2)	6.84 (d, J = 2) 6.80 (d, J = 8)	6.75 (d, J = 8)	6.75 $(dd, J_1 = 8, J_2 = 2)$	6.66 $(dd, J_1 = 8, J_2 = 2)$ 4.80 $(d, J = 6.5)$	2.52 $(dd, J_1 = 13, J_2 = 11)$	2.96 (dd, $J_1 = 13$ , $J_2 = 3$ ) 2.30 (m)	2 69 (m)	3.90-3.77*		$3.67 (dd, J_1 = 8.5, J_2 = 6.5)$	$3.96 (dd, J_1 = 8.5, J_2 = 6.5)$	3.83 ( <b>6H</b> , s)	7.37	7.47	1	1
12		23 7.01	75.0-70.7			3.05-2.21		ĺ	101-155		5.17 and 4.91 (1H $2 \times 4$ $J = 4$ and 6)	() Turk - , 'n , 7 (111)	3.84 (6H, br s)	7.34 and 7.33	7.43 and 7.34 3.71 and 3.30	1	5.53 and 5.19 (1H, $2 \times d$ , $J = 4$ and 6)
11	6.85 (d, J = 2)	7.01 (d, J = 2) 6.8-6.07*		6.82 $(dd, J_1 = 8, J_2 = 2)$	6.71 $(dd, J_1 = 8, J_2 = 2)$ 2.59 $(dd, J_1 = 13.5, J_2 = 11.5)$ 3.03 $(dd, J_1 = 13.5, J_2 = 3)$	2 94 (hr s)	1.97 (m)		3.61 (ddd, $J_1 = 11, J_2 \approx J_3 \approx 5$ )	3.76 (ddd, $J_1 = 11$ , $J_2 \approx J_3 \approx 3-4$ )	$3.45 (dd, J_1 = 11.5, J_2 = 6)$	$3.51 (dd, J_1 = 11.5, J_2 = 4)$	3.83 (6H, s)	7.35	7.33 3.61	$4.42 (dd, J_1 = 5, J_2 = 3)$	4.22 $(dd, J_1 = 6, J_2 = 4)$
10	6.79 (d, J = 2)	6.83 (d, J = 2) 6.76 (d, J = 8)	6.77 (d, J = 8)	6.66 $(dd, J_1 = 8, J_2 = 2)$	6.70 $(dd, J_1 = 8, J_2 = 2)$ 2.55* 2.86*	2.95 (d, J = 13.5)	3.15 (d, J = 13.5) 2.55*	I	$3.97 (dd, J_1 = 14, J_2 = 8)$	$4.02 (dd, J_1 = 14, J_2 = 8)$	í	- 52			.' 7.52 5.16	1	
	H-2	H-2' H-5	H-5,	9-H	H-6' H <sub>A</sub> -7	H <sub>A</sub> -7′	H <sub>B</sub> -7' H-8	H-8'	6-V Н	$^{\rm H_{B^-9}}$	H <sub>A</sub> -9′	H <sub>R</sub> -9'		ar-OH	ar-OH′ OH-8′	6-HO	ОН-9,

Coupling constants (J) in Hz. \*Different signals superimposed.

	10	11	12	14
C-1, C-1'	130.50, 126.27	132.38, 128.20	132.31, 132.14, 128.37, 127.97	132.4
C-2, C-2'	112.97, 111.73	113.28, 111.57	113.24, 113.16, 111.43	111.7
C-3, C-3'	146.87, 146.83	146.68*	146.83, 146.74, 146.70	146.6
C-4, C-4'	145.29, 144.59	144.75, 144.13	145.00, 144.92, 144.25	143.7
C-5, C-5'	114.74, 114.51	114.37*	114.60, 114.45	114.3
C-6, C-6'	123.37, 121.67	123.29, 121.78	123.33, 123.22, 121.43, 121.31	121.5
C-7	31.74	31.89	32.62, 31.94	35.8
C-7'	42.26	40.39	45.32, 38.73	
C-8	43.97	47.60	47.96, 45.61	43.7
C-8'	†	+	82.12, 79.08	
C-9	70.26	61.20	73.10, 71.12	60.5
C-9'	178.67	65.46	103.48, 101.14	
O-Me, O-Me'	56.10, 56.04	56.07	56.08, 56.01	55.7

Table 2. 13C NMR data of lignans (in CDCl<sub>3</sub>)

[mp 120–122°,  $[\alpha]_{0}^{20}$  – 118° (EtOH; c 0.35)], physicochemical data identical with 15 [20].

Carinol (11). C5 was separated over 250 g Sephadex LH 20 with MeOH-CHCl<sub>3</sub> (7:3) into five fractions (C5.1-C5.5). The soln of C5.4 in EtOAc was treated with 5% aq. Na<sub>2</sub>CO<sub>3</sub>, evaporated and rechromatographed over 30 g Si gel (CHCl<sub>3</sub>-MeOH, 19:1) to give 175 mg of an amorphous powder  $[\alpha]_D^{20} - 23^{\circ}$  (EtOH; c 0.4)], physicochemical data identical with 11 [19]. Acetylation (Ac<sub>2</sub>O-pyridine) converted the isolated material into a tetra-acetyl derivative, whose spectra were identical with published data for tetra-acetyl-11 [19].

Methylation of carissanol with MeOH-H+. To a soln of 20 mg 12 in 2 ml MeOH one drop of conc. H<sub>2</sub>SO<sub>4</sub> was added and the mixture stirred for 12 hr at room temp., then neutralized with K<sub>2</sub>CO<sub>3</sub>, filtered and evaporated. The residue was separated by chromatography over Si gel (CHCl<sub>3</sub>-MeOH, 49:1) into the epimeric methylation products I (amorphous, 6 mg = 29 %) and II (amorphous, 10.5 mg = 51 %). Epimeric product I:  $[\alpha]_D^{20}$  $-71.5^{\circ}$  (EtOH; c 0.2); MS: m/z (int.  $\geq 5\%$ ) 390.1692 C<sub>21</sub>H<sub>26</sub>O<sub>7</sub> (12, M+), 206 (5), 193 (30), 180 (5), 175 (9), 164 (5), 163 (15), 161 (6), 151 (7), 150 (15), 143 (6), 138 (41), 137 (100), 133 (6), 131 (8), 124 (9), 123 (7), 122 (11), 115 (5), 94 (9), 77 (8), 65 (5), 55 (14), 51 (5); UV  $\lambda_{\text{max}}^{\text{EiOH}}$  nm (log  $\varepsilon$ ): 228 (4.01), 280 (3.73); IR  $\nu_{\text{max}}^{\text{CHCl}_3}$  cm<sup>-1</sup>: 3550, 2945, 2850, 1612; <sup>1</sup>H NMR (CDCl<sub>3</sub>):  $\delta$ 6.86 (1H, d, J = 8 Hz, ar- $\underline{H}$ ), 6.84 (1H, d, J = 2 Hz, ar- $\underline{H}$ ), 6.81  $(1 \text{ H}, d, J = 8 \text{ Hz}, \text{ar-}\underline{H}), 6.74 (1 \text{ H}, dd, J_1 = 8 \text{ Hz}, J_2 = 2 \text{ Hz}, \text{ar-}\underline{H})$  $\underline{\mathbf{H}}$ ), 6.60 (1H, dd,  $J_1 = 8$  Hz,  $J_2 = 2$  Hz, ar- $\underline{\mathbf{H}}$ ), 6.54 (1H, d, J = 2 Hz, ar- $\underline{H}$ ), 5.54 (1 $H_{ex}$ , s, ar- $O\underline{H}$ ), 5.46 (1 $H_{ex}$ , s, ar- $O\underline{H}$ ), 4.71 (1H, s,  $C\underline{H} < O$ ), 3.89 (3H, s,  $OC\underline{H}_3$ ), 3.86 (3H, s,  $OC\underline{H}_3$ ), 3.82 (1H,  $d\bar{d}$ ,  $J_1 \approx J_2 \approx 8.5 \,\text{Hz}$ ,  $OCH_AH_B-CH$ ), 3.72  $(1H, J_1 \approx J_2 \approx 8.5 \text{ Hz}, \text{ OCH}_{\underline{AH}}_{\underline{B}}\text{-CH}), 3.37 (3H, s, \text{ OC}_{\underline{H}}_3),$ 3.04 (1H<sub>ex</sub>, s, OH), 2.81 (1H, d, J = 13.5 Hz, ar-CH<sub>A</sub>H<sub>B</sub>-C), 2.75 (1H, d, J = 13.5 Hz, ar-CH<sub>A</sub>H<sub>B</sub>-C), 2.75 (1H, dd,  $J_1$ = 14 Hz,  $J_2$  = 4 Hz, ar-C $\underline{H}_A$ H $_B$ -CH), 2.45 (1H, dd,  $J_1$  = 14 Hz,  $J_2 = 10.5 \text{ Hz}$ , ar-CH<sub>A</sub>H<sub>B</sub>-CH), 2.24 (1H, m). Epimeric product II:  $[\alpha]_D^{20} + 10.5^\circ$  (EtOH; c 1); MS: m/z (int.  $\ge 5\%$ ) 390.1699 C<sub>21</sub>H<sub>26</sub>O<sub>7</sub> (10, M<sup>+</sup>), 206 (6), 194 (6), 193 (33), 180 (5), 175 (9), 164 (5), 163 (8), 161 (7), 151 (8), 150 (16), 143 (7), 139 (6), 138 (42), 137 (100), 133 (7), 131 (5), 124 (10), 123 (6), 122 (11), 94 (9), 77 (8), 65 (5), 55 (15), 51 (5); UV  $\lambda_{\text{max}}^{\text{EIOH}}$  nm (log  $\varepsilon$ ): 228 (4.05), 279 (3.73); IR  $\nu_{\text{max}}^{\text{CHCl}_3}$  cm<sup>-1</sup>: 3540, 2935, 2845, 1608; <sup>1</sup>H NMR (CDCl<sub>3</sub>):  $\delta 6.88$  (1H, d, J = 8 Hz, ar- $\underline{H}$ ), 6.85 (1H, d, J= 8 Hz, ar-H), 6.83 (1H, br s, ar-H), 6.77 (1H, dd,  $J_1$  = 8 Hz,  $J_2$ = 2 Hz, ar- $\underline{H}$ ), 6.71 (1H, dd,  $J_1 = 8$  Hz,  $J_2 = 2$  Hz, ar- $\underline{H}$ ), 6.69 (1H, br s, ar- $\underline{H}$ ), 5.60 (1H<sub>ex</sub>, s, ar- $O\underline{H}$ ), 5.54 (1H<sub>ex</sub>, s, ar- $O\underline{H}$ ), 4.35 (1H, s,  $CH \subset O$ ), 3.97 (1H, dd,  $J_1 \approx J_2 \approx 8$  Hz, O- $CH_AH_B$ -CH), 3.89 (6H, s, 2 ×  $OC\underline{H}_3$ ), 3.80 (1H, dd,  $J_1 \approx J_2 \approx 8$  Hz, O- $CH_A\underline{H}_B$ -CH), 3.38 (3H, s,  $OC\underline{H}_3$ ), 3.05 (1H, d,  $J_1 = 14$  Hz, ar- $C\underline{H}_A\underline{H}_B$ -C), 2.87 (1H, dd,  $J_1 = 12.5$  Hz,  $J_2 = 4$  Hz, ar- $CH_A\underline{H}_B$ -CH), 2.67 (1H, d,  $J_1 = 14$  Hz, ar- $CH_A\underline{H}_B$ -C), 2.50-2.74 (2H, m), 1.71 (1H<sub>ex</sub>, s,  $O\underline{H}$ ).

Conversion of 12 into 11. LiAlH<sub>4</sub> (5 mg) was added to a soln of 10 mg 12 in 4 ml Et<sub>2</sub>O and stirred for 4 hr at room temp.; excess LiAlH<sub>4</sub> was destroyed with EtOAc, then 10 ml 5% aq. HCl was added and the aq. layer extracted  $\times$  5 with Et<sub>2</sub>O. Combined organic layers were washed, dried and purified by chromatography over 4 g Si gel (CHCl<sub>3</sub>-MeOH, 19:1) to give 7 mg (= 70%) of an amorphous powder  $[\alpha]_D^{20} - 21^\circ$  (EtOH; c 0.5) identical with 11.

Preparation of 12 from 10. Compound 10 (25 mg) was dissolved in 5 ml THF (freshly distilled over Na): after addition of 10 mg LiAlH<sub>4</sub> the reaction mixture was stirred for 4 hr at room temp. Work-up as described above and chromatography [5 g Si gel, CHCl<sub>3</sub>-MeOH (49:1)] afforded 8 mg 12 and 5 mg 11.

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<sup>\*</sup>Intensity significantly enhanced.

<sup>†</sup>Hidden under solvent signal.

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